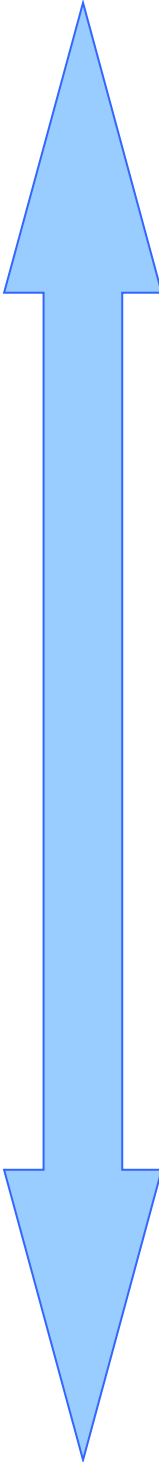


Semirreacción de reducción	ε° (V)	
$\text{Li}^+ + \text{e}^- \rightarrow \text{Li}_{(s)}$	-3,0401	+REDUCTOR 
$\text{K}^+ + \text{e}^- \rightarrow \text{K}_{(s)}$	-2,931	
$\text{Ca}^{2+} + 2\text{e}^- \rightarrow \text{Ca}_{(s)}$	-2,868	
$\text{Na}^+ + \text{e}^- \rightarrow \text{Na}_{(s)}$	-2,7144	
$\text{Mg}^{2+} + 2\text{e}^- \rightarrow \text{Mg}_{(s)}$	-2,3568	
$\text{Al}^{3+} + 3\text{e}^- \rightarrow \text{Al}_{(s)}$	-1,676	
$\text{Mn}^{2+} + 2\text{e}^- \rightarrow \text{Mn}_{(s)}$	-1,185	
$2\text{H}_2\text{O} + 2\text{e}^- \rightarrow \text{H}_{2(g)} + 2\text{OH}^-$	-0,8277	
$\text{Zn}^{2+} + 2\text{e}^- \rightarrow \text{Zn}_{(s)}$	-0,7628	
$\text{Cr}^{3+} + 3\text{e}^- \rightarrow \text{Cr}_{(s)}$	-0,74	
$\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}_{(s)}$	-0,440	
$\text{Cr}^{3+} + \text{e}^- \rightarrow \text{Cr}^{2+}_{(s)}$	-0,42	
$\text{Cd}^{2+} + 2\text{e}^- \rightarrow \text{Cd}_{(s)}$	-0,40	
$\text{Ni}^{2+} + 2\text{e}^- \rightarrow \text{Ni}_{(s)}$	-0,236	
$\text{Sn}^{2+} + 2\text{e}^- \rightarrow \text{Sn}_{(s)}$	-0,13	
$\text{Pb}^{2+} + 2\text{e}^- \rightarrow \text{Pb}_{(s)}$	-0,1266	
$2\text{H}^+ + 2\text{e}^- \rightarrow \text{H}_{2(g)}$	0,0000	
$\text{Cu}^{2+} + \text{e}^- \rightarrow \text{Cu}^+$	+0,159	
$\text{SO}_4^{2-} + 4\text{H}^+ + 2\text{e}^- \rightarrow \text{SO}_{2(g)} + 2\text{H}_2\text{O}$	+0,17	
$\text{Cu}^{2+} + 2\text{e}^- \rightarrow \text{Cu}_{(s)}$	+0,3394	
$\text{O}_{2(g)} + 2\text{H}_2\text{O} + 4\text{e}^- \rightarrow 4\text{OH}^-$	+0,414	
$\text{Cu}^+ + \text{e}^- \rightarrow \text{Cu}_{(s)}$	+0,5180	
$\text{I}_{2(s)} + 2\text{e}^- \rightarrow 2\text{I}^-$	+0,535	
$\text{MnO}_4^- + 2\text{H}_2\text{O} + 3\text{e}^- \rightarrow \text{MnO}_{2(s)} + 4\text{OH}^-$	+0,597	
$\text{Fe}^{3+} + \text{e}^- \rightarrow \text{Fe}^{2+}$	+0,769	
$\text{Ag}^+ + \text{e}^- \rightarrow \text{Ag}_{(s)}$	+0,7996	
$\text{NO}_3^- + 4\text{H}^+ + 3\text{e}^- \rightarrow \text{NO}_{(g)} + 2\text{H}_2\text{O}$	+0,96	
$\text{Br}_{2(l)} + 2\text{e}^- \rightarrow 2\text{Br}^-$	+1,066	
$\text{Br}_{2(aq)} + 2\text{e}^- \rightarrow 2\text{Br}^-$	+1,0873	
$2\text{IO}_3^- + 12\text{H}^+ + 10\text{e}^- \rightarrow \text{I}_{2(s)} + 6\text{H}_2\text{O}$	+1,2093	
$\text{O}_{2(g)} + 4\text{H}^+ + 4\text{e}^- \rightarrow 2\text{H}_2\text{O}$	+1,2288	
$\text{Cr}_2\text{O}_7^{2-} + 14\text{H}^+ + 6\text{e}^- \rightarrow 2\text{Cr}^{3+} + 7\text{H}_2\text{O}$	+1,33	
$\text{Cl}_{2(g)} + 2\text{e}^- \rightarrow 2\text{Cl}^-$	+1,3601	
$\beta\text{-PbO}_{2(s)} + 4\text{H}^+ + 2\text{e}^- \rightarrow \text{Pb}^{2+} + 2\text{H}_2\text{O}$	+1,458	
$\alpha\text{-PbO}_{2(s)} + 4\text{H}^+ + 2\text{e}^- \rightarrow \text{Pb}^{2+} + 2\text{H}_2\text{O}$	+1,468	
$\text{ClO}^- + 2\text{H}^+ + 2\text{e}^- \rightarrow \text{Cl}^- + \text{H}_2\text{O}$	+1,46	
$\text{MnO}_4^- + 8\text{H}^+ + 5\text{e}^- \rightarrow \text{Mn}^{2+} + 4\text{H}_2\text{O}$	+1,5119	
$\text{Au}^{3+} + 3\text{e}^- \rightarrow \text{Au}_{(s)}$	+1,52	
$\text{H}_2\text{O}_{2(l)} + 2\text{H}^+ + 2\text{e}^- \rightarrow 2\text{H}_2\text{O}$	+1,78	
$\text{Au}^+ + \text{e}^- \rightarrow \text{Au}_{(s)}$	+1,83	
$\text{F}_{2(g)} + 2\text{e}^- \rightarrow 2\text{F}^-$	+2,890	+OXIDANTE